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Polyphenols rich fraction from *Geoffroea decorticans* fruits flour affects key enzymes involved in metabolic syndrome, oxidative stress and inflammatory process

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2 **in metabolic syndrome, oxidative stress and inflammatory process**

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13 **Running title: Polyphenols and biological activity of *chañar* fruits flour**

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27 **Abstract**

28 *Geoffroea decorticans* (chañar), is widely distributed throughout Northwestern Argentina. Its fruit
29 is consumed as flour, arropo or hydroalcoholic beverage. The chañar fruits flour was obtained and
30 39 phenolic compounds were tentatively identified by HPLC-MS/MSⁿ. The compounds comprised
31 caffeic acid glycosides, simple phenolics (protocatechuic acid and vanillic acid), a glycoside of
32 vanillic acid, *p*-coumaric acid and its phenethyl ester as well as free and glycosylated flavonoids.
33 The polyphenols enriched extract with and without gastroduodenal digestion inhibited enzymes
34 associated with metabolic syndrome, including α -amylase, α -glucosidase, lipase and hydroxyl
35 methyl glutaryl CoA reductase. The polyphenolic extract exhibited antioxidant activity by different
36 mechanisms and inhibited the pro-inflammatory enzymes (cicloxygenase, lipooxygenase and
37 phospholipase A₂). The polyphenolic extract did not showed mutagenic effect by Ames test against
38 *Salmonella typhimurium* TA98 and TA100 strains.

39 These findings add evidence that chañar fruit flour may be considered a functional food with
40 preventive properties against diseases associated with oxidative stress, inflammatory mediators and
41 metabolic syndrome.

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45 **Keywords:** *Geoffroea decorticans*; chañar flour; polyphenolic compounds; metabolic syndrome;
46 antioxidant activity; anti-inflammatory activity; genotoxicity

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50 1. Introduction

51 The metabolic syndrome (MetS) is a state of insulin resistance, oxidative stress and chronic
52 inflammation that affect 25% of the world population (Prasad, Ryan, Celzo, & Stapleton, 2012).
53 MetS is characterized by the presence of at least three of the cardiovascular risk factors: obesity,
54 excessive visceral fat storage, dyslipidemia, hypertension, hyperglycemia or type 2 diabetes.
55 Treatments for MetS are mainly based on therapeutic lifestyle changes, often accompanied by
56 pharmacological treatments of MetS-related factors (Prasad et al., 2012). Reduction of postprandial
57 hyperglycemia by inhibition of enzymes involved in carbohydrate metabolism (α -glucosidase and
58 α -amylase), inhibition of lipolytic enzymes including lipase, inhibition of oxidative stress and the
59 delay of inflammatory process are the most common therapeutic approach to treat MetS. Plant-
60 derived foods represent a natural source of phytochemical positively associated with the prevention
61 and regression of MetS clinical manifestations. In particular, plant antioxidants are known to reduce
62 the oxidative stress and inflammatory process associated with obesity and cardiovascular alterations
63 (Abete, Goyenechea, Zulet, & Martinez, 2011). The evidence suggests that a diet enriched in fibers
64 and low in saturated fats, together with increased daily physical exercise, can reduce the incidence
65 of diabetes in people with impaired fasting glucose tolerance by almost 60% (Wagh & Stone,
66 2004).

67 The tree *Geoffroea decorticans* (Gill.ex Hook. et Arn.) Burk (Fabaceae), known as chañar, occurs
68 in the native forests of the Gran Chaco region (phytogeographical regions of “Parque chaqueño”,
69 “Monte” and “Espinal”) in Argentina as well as in the Paraguayan and Bolivian Chaco and northern
70 Chile (Scarpa, 2009). It is a xerophytic tree that forms extensive uniform colonies due to
71 reproduction by gemmiferous root. The ripe fruits (drupes) were consumed since pre-colonial
72 times. At present, they are still part of the diets of aboriginal communities like the Wichis, Toba and

73 Chorotes (Arena & Scarpa, 2007). In summer time, each tree produces between 20 and 30 kg of
74 orange brown colored fruits, which fall from the tree when mature (Figure 1). Under dry conditions,
75 the fruits can be stored for one year before consumption. The fruits and seeds are used raw, roasted,
76 boiled and/or fermented (beverage). The chañar flour was used as ingredient for soups and to make
77 so-called chañar breads. The other popular product made of chañar fruits is a sweet jelly-like syrup
78 called arrope. The arrope is recommended as a sweet for desserts and cocktails as well as the best
79 cough syrup in traditional medicine. Chañar fruits flour and arrope can be used as functional food
80 due to their high level of sugar, fiber and polyphenolic compounds (Costamagna, Ordoñez,
81 Zampini, Sayago & Isla, 2013). The antinociceptive action and antioxidant activity of chañar fruits
82 and arrope was reported (Costamagna et al., 2013; Reynoso, Vera, Aristimuño, Daud & Sánchez
83 Riera, 2013).

84 Chañar is relevant in the native Argentine forests and is often found associated with algarrobo
85 (*Prosopis*) and mistol (*Zyziphus*) trees. The collection and processing of chañar fruits, that was
86 important in the past, is now disappearing due to the clearance of the native forests for agriculture,
87 mainly to produce transgenic soybean for export. Therefore, studies that add value to this species
88 are important to promote their propagation, conservation and sustainable management in arid areas.
89 The aim of the present study was to assess the polyphenolic profile of chañar fruits flour and to
90 evaluate the efficacy of polyphenolic component against enzymes relevant in hyperglycemia,
91 dyslipidemia, oxidative stress and inflammatory process related with metabolic syndrome.

92 **2. Materials and Methods**

93 **2.1. Reagent and standards**

94 All chemicals and reagent used were either analytical reagent or HPLC grade. The chemicals,
95 standard compounds and enzymes purchased from Sigma Aldrich Co., St. Louis, USA were: p-

96 nitrophenyl α -D- glucopyranoside (pNPG, BCBN2675V); p-nitrophenyl palmitate
97 (pNPP109K5200); hydroxy methylglutaril CoA (HMG-CoA, L 112M4102V); linoleic acid
98 1393368) ; bovine serum albumin, (BSA,60F-0270); β -carotene (N° 117H2511); thiobarbituric acid
99 (N°034K06055); NADH (D00079245); dinitrophenyl hydrazine (DNPH) (S16313-065); ascorbic
100 acid (A596050-81-7); quercetin (Q4951); ABTS (1392678); lipoxygenase (LOX-1, L2630);
101 secretory phospholipase A2 (sPLA2, 0424023-1); α -glucosidase (N° 117F8205); pepsin
102 (128K7354V) and pancreatin (110M1429V).

103 COX-2 enzyme immune assay EIA (Kit N° 560131) was obtained from Cayman Chemical, MI,
104 USA.

105 Lipase pancreatic (N° BCR693); α -amylase (N° 67688), ABTS (N° 11557) and BHT (B1215000)
106 were purchased in Fluka.

107 **2.2. Plant material**

108 Fruits of *Geoffroea decorticans* were collected in January 2011 and January 2012 in Departamento
109 Fernández, Provincia de Santiago del Estero, Argentina. The fruits were lyophilized and the pulp
110 was ground to powder (Helix mill, Metvisa, Argentina). Then, the chañar flour was passed through
111 a 4 mm mesh sieve. The tree growing in the natural habitat and details of the fruits are presented in
112 Figure 1.

113 **2.3. Extraction and phenolic content determination**

114 The flour sample (630 g) was extracted four times with methanol:water 70:30 v/v in a 1:2
115 w/v ratio in an ultrasonic bath for 30 min at 25°C. Then, the combined extract was concentrated
116 under reduced pressure and then lyophilized. The lyophilized extract was suspended in
117 distilled water (acidified with HCl to pH 2) and mixed with Amberlite-XAD 7 resin (500 g) for 1
118 hour under stirring. The resin was filtered on a Büchner funnel, washed three times with the

119 acidified water and the polyphenols bound to the resin were desorbed with methanol (MeOH). The
120 phenolic-enriched MeOH solution was concentrated under vacuum to afford a crude MeOH extract
121 that was stored at -20°C until use. Total polyphenols were determined by the Folin–Ciocalteu
122 method (Singleton, Orthofer, Lamuela-Raventos, 1999). Results were expressed as mg gallic acid
123 equivalents (g GAE)/100 g dry weight (DW) of soluble principles. The flavonoids content was
124 determined according to Popova, Silici, Kaftanoghu & Bankova, (2005) and expressed as µg
125 quercetin equivalents per mg of dry weight (g QE/ 100 g DW).

126 **2.4. Enzyme inhibition and antioxidant assays**

127 **2.4.1. Inhibitory activity of enzymes related to metabolic syndrome**

128 The inhibitory effect on enzyme related to metabolic syndrome was determined using polyphenolic
129 extract with and without gastroduodenal digestion (GD). The polyphenolic extract was submitted to
130 GD into three categories: salivary, gastric and duodenal digestion according to Tenore et al., 2015
131 with minor modifications. Briefly: For the salivary digestion, the extract (4 mg GAE) were mixed
132 with 6 mL of saliva at pH 6.8 (KCl [89.6 g/L], KSCN [20 g/L], NaH₂PO₄ [88.8 g/L], Na₂SO₄ [57.0
133 g/L], NaCl [175.3 g/L], NaHCO₃ [84.7 g/L], urea [25.0 g/L] and α-amylase [48.3 mg/mL]). The
134 mixture was incubated for 3 min at 37°C. Then, for the gastric digestion, pepsin (14,800 U)
135 dissolved in HCl 0.1 M was added, pH was adjusted to 2 and the mixture was incubated at 37°C
136 during 2 h. For pancreatic (duodenal) digestion, the pH was adjusted to 6.5 with NaHCO₃ (0.5 M).
137 Then, pancreatin (8 mg/mL) and bile salts (50 mg/mL) (1:1, v/v), dissolved in water (20 mL), was
138 added and the mixture was incubated at 37 °C for 2 h. After digestions, polyphenols were extracted
139 with ethyl acetate and the organic phase was taken to dryness and resuspended in DMSO (2 mg
140 GAE/mL).

141 **2.4.1.1. α-Glucosidase inhibition**

142 The inhibition of α -glucosidase was determined using *p*-nitrophenyl α -D-glucopyranoside as
143 substrate. The reaction mixture contained 160 μ l of 0.1M sodium phosphate buffer (pH 6.9), 5 μ L
144 of enzyme (5.46 U/mL) and polyphenolic extract (0.17-1.36 μ g/mL). After pre-incubation of the
145 reaction mixture on ice for 5 min, the enzyme reaction was started by adding 5 μ L of 25 mM *p*-
146 nitrophenyl α -D-glucopyranoside into this mixture. The reaction was incubated 15 min at 37 $^{\circ}$ C.
147 Then, 80 μ L of 0.2 M sodium carbonate was added. The absorbance was measured at 405 nm in a
148 microplate reader (BiotekELx808). Enzyme inhibition was calculated using the following equation:

149
$$\% \text{ inhibition} = (A_0 - A_s) / A_0 \times 100$$

150 Where A_0 is the absorbance of the control (blank, without extract) and A_s is the absorbance in
151 presence of the extract. IC₅₀ values denote the μ g GAE/mL required to inhibit the enzyme by 50%.

152 **2.4.1.2. α -Amylase inhibition**

153 The α -amylase inhibitory activity using starch as substrate was assayed using Amilokit $\text{\textcircled{R}}$
154 (Wiener Lab Group, Rosario, Argentina). The reaction mixture contained 800 μ L of 0.01 M sodium
155 phosphate buffer (pH 7.4), 5 μ L of enzyme and polyphenolic extract (28-146 μ g GAE/mL). After
156 pre-incubation of the reaction mixture on ice for 5 min, the enzyme reaction was started by adding
157 500 μ L of reagent A (substrate) into the reaction mixture. Then, the reaction was incubated at 37 $^{\circ}$ C
158 for 7 min. After that, 500 μ L of reagent B (iodine solution) was added for color development and
159 the reaction mixture was taken to a final volume of 5.3 mL with water. The absorbance was
160 measured at 640 nm in a spectrophotometer (UV2400 PC). Enzyme inhibition was calculated using
161 the following equation:

162
$$\% \text{ inhibition} = (A_0 - A_s) / A_0 \times 100$$

163 Where A_0 is the absorbance of the control (blank, without extract) and A_s is the absorbance in
164 presence of the extract. IC₅₀ values denote the μ g GAE/mL required to inhibit the enzyme by 50%.

165 **2.4.1.3. Lipase inhibition**

166 Lipase activity was assayed by measuring the enzymatic hydrolysis of *p*-nitrophenyl palmitate to *p*-
167 nitrophenol in a microplate reader (BiotekELx808) at 400 nm. Lipase solution (1.0 mg/mL) was
168 mixed with the polyphenolic extract (final concentration between 0.2 and 8.71 µg/mL) and pre-
169 incubated on ice for 5 min. The reaction mixture for standard assay contained 330 µL of sodium
170 phosphate buffer 0.1 M (pH 7) supplemented with 0.6% (w/v) Triton X-100 and 0.15% (w/v) arabic
171 gum, and 20 µL of 10 mM *p*-nitrophenyl palmitate. The enzyme reaction was started by adding 50
172 µL of the lipase/ polyphenolic extract solution into the reaction mixture, and incubated at 37 °C for
173 20 min. Enzyme inhibition was calculated using the following equation:

$$174 \quad \% \text{ inhibition} = (A_0 - A_s) / A_0 \times 100$$

175 Where A_0 is the absorbance of the control (blank, without extract) and A_s is the absorbance in
176 presence of the extract. IC_{50} values denote the µg GAE/mL required to inhibit the enzyme by 50%.

177 **2.4.1.4. HMG-CoA Reductase inhibition**

178 The HMG-CoA Reductase KIT Assay was used. The reaction mixture contained extract (0.7 to 13
179 µg GAE/mL) or 1µL of parvastatin (as inhibitor), 1x buffer to complete 182 µL, 4 µL of NADPH,
180 12 µL of HMG-CoA and 2 µL of HMG-CoA reductase. The reaction was incubated at 37°C during
181 10 min. Readings were taken every minute for 10 minutes at 340 nm in a microplate reader. IC_{50}
182 values denote the µg GAE/mL required to inhibit the enzyme by 50%.

183 **2.4.2. Inhibition of pro-inflammatory enzymes**

184 **2.4.2.1. Lipoxygenase**

185 Lipoxygenase (LOX) activity was determined using a spectrophotometric method based on the
186 enzymatic oxidation of linoleic acid to the corresponding hydroperoxide. The reaction mixture
187 contained substrate (50 µM linoleic acid in 0.2 M borate buffer pH 9), enzyme (0.9 nM soy LOX-1)

188 and different concentrations of polyphenolic extract (100-600 $\mu\text{g GAE/mL}$). The assay to obtain the
189 100% of LOX activity was performed with DMSO as solvent control. Absorption at 234 nm was
190 recorded as a function of time for 3 min. The polyphenolic concentration causing 50% inhibition of
191 hydroperoxide-release (IC_{50}) was calculated from the concentration–inhibition response curve by
192 regression analysis. The extinction coefficient of $25 \text{ mM}^{-1} \text{ cm}^{-1}$ was used for hydroperoxide
193 quantification. Naproxen was used as a reference anti-inflammatory compound.

194 **2.4.2.2. Cyclooxygenase**

195 The ability of the extracts and/or fractions to inhibit the conversion of arachidonic acid to
196 prostaglandin (PG) by human recombinant COX-2 was determined by enzyme immune assay, EIA.
197 Cyclooxygenase catalyzes the first step in the biosynthesis of arachidonic acid to PGH₂. PGF₂ α
198 produced from PGH₂ by reduction with stannous chloride was measured by EIA in a microplate
199 reader (BiotekELx 808). The assays were performed in presence of 100-600 $\mu\text{g GAE/mL}$ of
200 polyphenolic extract or commercial anti-inflammatory drugs (nimesulide selective for COX-2). The
201 assay to obtain the 100% of COX-2 activity was performed with and without DMSO as solvent
202 control. Enzyme control was performed with inactivated enzymes by boiling during 3 min.
203 Detection limit was 29 $\mu\text{g PG/mL}$. The polyphenolic concentration causing 50% inhibition of
204 enzyme (IC_{50}) was calculated from the concentration–inhibition response curve by regression
205 analysis.

206 **2.4.2.3. Phospholipase A₂**

207 Secretory phospholipase A₂ activity was determined using 1,2-diheptanoylthio-
208 glycerophosphocholine (1,2 dHGPC) and Triton X-100 as substrates. The mixture contained 50 μL
209 of buffer Tris–HCl (10 mM, pH 8), 10 μL of 5,5'- dithiobis-2-nitrobenzoic acid (DTNB) (10 mM),
210 10 μL of PLA₂ enzyme (1 mg/mL) and 100-600 $\mu\text{g GAE/mL}$ of the assayed samples dissolved in

211 DMSO or commercial anti-inflammatory drug (acetylsalicylic acid). The reaction was initiated by
212 the addition of 150 μ L of 1,2 dHGPC (1.66 mM) and maintained during 20 min at 25 °C. The
213 absorbance was read at 414 nm during 20 min every 2 min in a microplate reader (BiotekELx808).
214 The polyphenolic concentration causing 50% inhibition of enzyme (IC_{50}) was calculated from the
215 concentration–inhibition response curve by regression analysis.

216 **2.4.3 Antioxidant activity**

217 **2.4.3.1. ABTS radical scavenging**

218 The assay was carried out by the improved ABTS^{•+} method as described by Re, Pellegrini,
219 Proteggente, Pannala, Yang, & Rice-Evans (1999). ABTS^{•+} was generated by reacting 7 mM ABTS
220 and 2.45 mM potassium persulfate after incubation at room temperature (23 °C) in the dark for 16 h.

221 **2.4.3.1.1. Autographic assay on TLC**

222 An aliquot of each fraction obtained from total polyphenolic extract after fractionation by
223 Sephadex LH 20 was placed on Silica gel F254 plates (4x4 cm). The plates were developed with
224 (toluene: ethyl acetate: formic acid; 4:2:1, v:v:v) as solvent system. After, 3 mL of soft medium
225 (agar 0.9 %) containing 1mL ABTS^{•+} solution was distributed on TLC plate (Zampini, Ordoñez, &
226 Isla, 2010) After solidification, the plate was incubated at room temperature for 1 min in the dark.
227 The antioxidant activity appeared as clear spots against a dark green-blue background.

228 **2.4.3.1.2. Total antioxidant capacity by spectrophotometric assay**

229 ABTS^{•+} solution (1mL; absorbance of 0.7 ± 0.02 at 734 nm) was added to the sample (1.35 to
230 12.5 μ g GAE/mL) and mixed thoroughly. Absorbance was recorded at 734 nm after 6 min. The
231 percentage of inhibition was measured by the following formula:

$$232 \quad \% \text{ inhibition} = (A_0 - A_s) / A_0 \times 100$$

233 Where A_0 is the absorbance of the control (blank, without extract) and A_s is the absorbance in
234 presence of the extract. Results are presented as SC_{50} values in $\mu\text{g GAE/mL}$ required to scavenge
235 50% ABTS free radicals.

236 **2.4.3.2. Hydroxyl radical scavenging**

237 The reaction mixture contained 50 μL of 10.4 mM 2-deoxy-D-ribose and 100 μL of 50 μM FeCl_3
238 and extract (0.29-2.39 $\mu\text{g GAE/mL}$). The reaction was carried out with and without 100 μL of 52
239 μM EDTA. To start the Fenton reaction, 50 μL of 10 mM H_2O_2 and 50 μL of 1.0 mM ascorbic acid
240 were added. The mixture was incubated at 37°C for 60 min. Then, 500 μL of 2-thiobarbituric acid
241 (1%, w/v) dissolved in trichloroacetic acid (3%, w/v) was added. The tubes were heated at 100°C
242 for 20 min. The absorbance was measured at 532 nm. Reaction mixtures without the polyphenolic
243 extract were used as positive control (100% MDA). The negative control contained the full reaction
244 mixture without 2-deoxy-D-ribose. A modification was carried out without the addition of ascorbic
245 acid. The negative control contained the full reaction mixture without 2-deoxy-D-ribose. The
246 positive control was the $\text{H}_2\text{O}_2/\text{Fe}^{3+}/\text{ascorbic acid}$ system mixture lacking the extract (100% MDA).
247 Results are presented as SC_{50} values in $\mu\text{g GAE/mL}$ required to inhibit by 50% the degradation of
248 2-deoxy-D-ribose.

249 **2.4.3.3. Hydrogen peroxide scavenging**

250 A solution of hydrogen peroxide (4 mM) was prepared in phosphate buffer (PBS, pH 7.4).
251 Hydrogen peroxide concentration was determined spectrophotometrically from absorption at 230
252 nm using the molar absorptivity 81 $\text{M}^{-1}\cdot\text{cm}^{-1}$. Samples (7-82.7 $\mu\text{g GAE/mL}$) were added to the
253 hydrogen peroxide solution (0.6 mL). Absorbance of hydrogen peroxide at 230 nm was determined
254 10 min later against a blank solution containing extracts without hydrogen peroxide. Results are
255 presented as SC_{50} values in $\mu\text{g GAE/mL}$ required to inhibit by 50% the degradation of H_2O_2

256 **2.4.3.4. Protection of protein against oxidative damage**

257 Bovine serum albumin (BSA, 40 mg/mL) in 10 mM sodium phosphate buffer pH 7 was mixed with
258 1 mL of 1.5 mM FeSO₄, 1 mL H₂O₂ 3 mM and different concentrations of the samples. Then, 400
259 μL of 7 mM DNPH in 2 M HCl was added. The protein was precipitated by adding 500 μL of
260 trichloroacetic acid (4% w/v). The pellet collected by centrifuging for 5 min at 14,000 x g was re-
261 dispersed in ethanol/ethyl acetate (1:1, v/v) in order to remove unreacted DNPH. Absorbance at 378
262 nm was read and carbonyl concentration was calculated. Results are reported as PC₅₀ values in μg
263 GAE/mL required to inhibit by 50% the degradation of BSA.

264 **2.4.3.5. Protection of lipid against oxidative damage: β-Carotene bleaching**

265 Antioxidant activity was determined according to the β-carotene bleaching method following the
266 procedure described by Ordoñez, Gomez, Vattuone, & Isla (2006). The initial absorbance at 470 nm
267 was registered at time zero (t₀) and during 120 min. Antioxidant activity (AA%) was calculated as
268 the percent inhibition relative to control using the following equation:

$$269 \quad AA\% = [(R_{\text{control}} - R_{\text{sample}}) / R_{\text{control}}] \times 100$$

270 Where R_{control} and R_{sample} are the bleaching rates of β-carotene in the reactant mix without
271 antioxidant and in presence of the extracts, respectively. SC₅₀ values denote the μg
272 GAE/mL required to inhibit β-carotene bleaching by 50%.

273 **2.5. Mutagenicity**

274 **2.5.1. Salmonella mutagenicity assay**

275 The mutagenic effect of chañar fruits flour extracts was evaluated on two *S. typhimurium* strains
276 (TA98 and TA100). The plate incorporation assay was performed according to Maron & Ames
277 (1983), by adding 0.1 mL of the overnight bacterial culture, 0.1 mL of chañar polyphenolic extracts
278 at different concentrations (25-100 μg GAE/plate) and 2 mL of top agar on minimal agar. The

279 plates were then incubated at 37 °C for 48 h. After incubation, his⁺ revertant colonies were counted
280 and compared to the number of revertant colonies in the controls.

281 For the experiment with S9 metabolic activation, the S9 mixture was freshly prepared before the
282 assay and kept on ice until needed. The S9 mixture consisted of S9 fraction (Moltox – Molecular
283 Toxicology Inc., USA) containing NADP (Maron & Ames, 1983). The experiment was repeated as
284 above and the only difference was that 500 µL of S9 was added in place of the phosphate. After 72
285 h of incubation at 37 °C His⁺ revertants were counted. The positive controls employed were 4-nitro-
286 *O*-phenylenediamine (4-NPD; Aldrich Chemical Co.), (10 µg/plate) and 2-aminofluorene (2-AF;
287 Merck) (10 µg/ plate). Solvent control was carried out adding 100 µL DMSO/plate. An extract was
288 considered mutagenic when the mean number of revertants was double or greater than two times
289 that of the negative control. Three plates per experiment and two separate experiments were used
290 for each concentration tested and for positive and negative controls.

291 **2.6. Fractionation of the polyphenolic extract**

292 A sample of the polyphenolic extract (1.6 g) obtained from chañar fruits flour was dissolved in
293 MeOH and permeated on Sephadex LH-20. MeOH was used as mobile phase. The eluates with
294 similar TLC profiles (toluene: ethyl acetate: formic acid; 4:2:1, v:v:v), revealed with diphenylboric
295 acid ethanolamine complex, were pooled in four major fractions. All of them were taken to dryness
296 and kept at 4°C for further use. The yields were as follow. F-I: 151 mg; F-II: 1100 mg; F-III: 200
297 mg; F-IV: 64.7 mg. The fractions F-III and F-IV showed the highest antioxidant activities by
298 autographic assays. The FIII and FIV were analyzed by HPLC-DAD y HPLC-MS/MSⁿ. Then, FIII
299 and FIV were re-chromatographed on Sephadex LH 20 and eluted with Metanol:H₂O, (8:2, v:v)
300 Nine subfractions (SF-1 to SF-9) were obtained from FIII and eight sub-fractions (SF-1 to SF-8)

301 were obtained from FIV. All fractions and sub-fractions were dried by evaporation and lyophilized.

302 Dry extracts were kept at 4°C for further experimental use.

303 **2.7. Identification of phenolics**

304 **2.7.1. HPLC-DAD analysis**

305 The most active fractions of the polyphenolic extract (F-III and F-IV) and subfractions obtained
306 from them were analyzed by HPLC coupled to a diode array detector to set the conditions for
307 HPLC-DAD-MS/MS studies. The HPLC system used for DAD analysis was a Shimadzu
308 (Shimadzu Corporation, Kyoto, Japan) equipment consisting of a LC-20AT pump, a SPD-M20A
309 UV diode array detector, CTO-20AC column oven and a LabSolution software. A MultoHigh 100
310 RP 18-5 μ (250x4.6 mm) column (CS-Chromatographie Service GmbH- Germany) maintained at 35
311 °C was used. The samples were dissolved in 1 mL MeOH, filtered through a 0.45 μ m PTFE filter
312 (Waters) and submitted to HPLC-DAD and HPLC-ESI-MS/MS analysis. The compounds occurring
313 in the mixtures were monitored at 254 and 330 nm, and UV spectra were recorded from 200 to 600
314 nm for peak characterization. The HPLC analyses were performed using a linear gradient solvent
315 system consisting of 0.1% acetic acid in water (A) and methanol 0.1% acetic acid as follow: 90%
316 A to 43% A over 45 min, followed by 43% A to 0% A from 45 to 60 min, 0% A to 0% A from 60
317 to 5 min. The flow rate was 0.5 mL/min and the volume injected was 20 μ L.

318 **2.7.2. Identification of phenolics by HPLC-ESI-MS/MS**

319 Data were recorded on a HPLC-ESI-MS/MS system which consisted of the HPLC HP1100 (Agilent
320 Technologies Inc, CA-USA) connected through a split to the mass spectrometer Esquire 4000 Ion
321 Trap LC/MS(n) system (Bruker Daltonik GmbH, Germany). Ionization was performed at 3000 V
322 assisted by nitrogen as nebulizing gas at 24 psi and as drying gas at 365°C and a flow rate of 6

323 L/min. Negative ions were detected using full scan (m/z 20-2200) and normal resolution (scan
324 speed 10,300 $m/z/s$; peak with 0.6 FWHM/ m/z). The trap parameters were set in ion charge control
325 (ICC) using manufacturer default parameters, and maximum accumulation time of 200 ms.
326 Collision induced dissociation (CID) was performed by collisions with helium background gas
327 present in the trap and automatically controlled through SmartFrag option.

328 **2.8. Statistical analysis**

329 Sampling and analyses were performed in triplicate, and the data are presented as mean \pm standard
330 deviation (S.D.). The correlation between the main polyphenolic compounds content of the extracts,
331 and the biological activities was analyzed by the Pearson test correlation coefficients with 95%
332 confidence. Statistical analysis was performed by one way ANOVA followed by Tukey's multiple
333 comparison test ($p < 0.05$). All statistical analyses were carried out using the Infostat software.

334 **3. Results and discussion**

335 The macronutrient and phytochemical composition of chañar fruits flour was previously reported
336 (Costamagna et al., 2013). The carbohydrates were the major component (19.75 %) followed by
337 proteins (5%). Furthermore, the chañar fruits flour could be considered a source of dietary fiber
338 (12%), of potassium and polyphenolic compounds (1.5 %) (Costamagna et al., 2013). In the present
339 paper, a polyphenols-enriched hydroalcoholic extract was obtained from chañar fruits flour. The
340 polyphenolic compounds content and flavonoids of the extract were 12.5 ± 1.0 g GAE/100 g dry
341 weight of soluble principle and 2.0 ± 0.2 g QE/100 g dry weight of soluble principle, respectively.

342 The aim of the present study was to determine the efficacy of polyphenolic component against
343 enzymes involved in hyperglycemia, dyslipidemia, and inflammatory process related with
344 metabolic syndrome and oxidative stress.

345 **3.1. Enzyme inhibition**

346 The metabolic syndrome is a metabolic disorder of multiple etiologies characterized by chronic
347 hyperglycemia and disturbances of carbohydrate and fat metabolisms. The activity of the chañar
348 fruits flour polyphenolic extract was assessed towards enzymes associated with metabolic
349 syndrome, including α -glucosidase and α -amylase, pancreatic lipase and HMGCoA reductase, pro-
350 inflammatory enzymes (COX-1, COX-2, LOX, sPLA₂) and as antioxidant. The effect of the chañar
351 polyphenols on the different enzymes is summarized in Table 1. Furthermore, the fraction was
352 evaluated for a possible mutagenic effect by the Ames test.

353 **3.1.1. α -Glucosidase and α -amylase inhibition**

354 One therapeutic approach to decrease postprandial hyperglycemia is to suppress the production
355 and/or absorption of glucose from the gastrointestinal tract through inhibition of the enzymes α -
356 amylase or α -glucosidase (Abete et al., 2011). Hypoglycaemic agents used in clinical practice, such
357 as acarbose, competitively inhibit α -glucosidase in the brush border of the small intestine, which
358 consequently delay the hydrolysis of carbohydrates and alleviate postprandial hyperglycemia.
359 However, the continuous administration of these agents may cause several adverse effects, such as
360 diarrhea, abdominal discomfort, flatulence, and hepatotoxicity. Therefore, α -glucosidase novel
361 natural inhibitors are necessary given the therapeutic challenge of type II diabetes mellitus. Phenolic
362 compounds of chañar fruits flour may freely interact with enzymes present in the digestive tract
363 modulating their activity (Abete et al., 2011; Xiao, Chen, & Cao, 2014). The inhibitory activity of
364 polyphenolic extract obtained from chañar flour towards α -amylase was low (IC₅₀ values of 25 μ g
365 GAE/mL). However, this extract was very active against α -glucosidase (IC₅₀ = 0.68 μ g GAE/mL)
366 with a relation dose-response with polyphenolic concentration until IC₅₀ values (R^2 = 0.95),
367 presenting better effect than the reference compound acarbose (IC₅₀ = 25 μ g/mL), Table 1. These
368 results suggest that chañar fruit flour polyphenols might be able to reduce glucose

369 uptake/absorption. The selective inhibition of α -glucosidase is the preferred effect for plant extracts
370 to control glucose uptake. Simultaneous inhibition of both enzymes would result in abnormal
371 bacterial fermentation in the colon due to the presence of undigested carbohydrates. Other plants,
372 many of them used traditionally to control diabetes or hyperglycaemia were reported to exert strong
373 inhibition of α -glucosidase and moderate or negligible effect on α -amylase activity (Ranilla, Kwon,
374 Apostolidis, & Shetty, 2010). The activity of polyphenolic extract from chañar fruits flour on α -
375 glucosidase was higher than the activity of hydroalcoholic blueberry extract (Boath, Stewart,
376 McDougall, 2012), pomegranates (Medjakovic & Jungbauer, 2013) and maqui (Rubilar, et al.,
377 2011) fruits. In this sense, chañar fruits flour may offer dietary coadjuvants (therapeutic
378 complements) to control hyperglycemia in diabetic patients. However, further evaluation of their in
379 vivo hypoglycemic activity is necessary to verify these beneficial effects.

380 **3.1.2. Pancreatic lipase and HMGCoA reductase inhibition**

381 The inhibition of pancreatic lipase, which splits triacylglycerols into absorbable monoacylglycerol
382 and fatty acids, is the main prescribed treatment for weight management and obesity in developed
383 countries. Orlistat, one of the two clinically approved drugs for obesity treatment, has been shown
384 to act through inhibition of pancreatic lipase. In order to find alternative natural sources for obesity
385 prevention and treatment, we evaluated the chañar polyphenolic extract on lipase activity. Results
386 are showed in Table 1. The inhibitory activity of chañar polyphenolic extract on lipase (IC_{50} : 4 μ g
387 GAE/mL) was higher than that reported for white and green tea polyphenols (Gondoin, Grussu,
388 Stewart, & McDougall, 2010). Polyphenols from common plant foodstuffs such as tea, soybean,
389 mate tea, peanut, or grapevine have been reported as pancreatic lipase inhibitors (Garza, Milagro,
390 Boque, Campion, & Martinez, 2011). A positive correlation between lipase inhibitory activity of

391 chañar flour and total phenolics ($R^2=0.95$) was demonstrated as well as in several dietary
392 supplements and fruits (Garza et al., 2011).

393 Hypercholesterolemia and cardiovascular disease are major health problems. One approach to
394 reduce hypercholesterolemia is to use medicines that inhibit the enzymes essential for cholesterol
395 biosynthesis. HMGCoA reductase catalyzes the rate-limiting step in cholesterol biosynthesis.
396 Inhibition of cholesterol synthesis lowers the hepatocyte cholesterol content and increase expression
397 of low density lipoprotein cholesterol (LDL-c) receptors, responsible for LDL-c uptake via
398 receptor-mediated endocytosis, and consequently LDL-c is rapidly cleared from the bloodstream.
399 Statins (lovastatin, simvastatin, pravastatin, fluvastatin, atorvastatin, pitavastatin) are HMG-CoA
400 reductase inhibitors which are highly effective therapeutic agents for the treatment of
401 hypercholesterolemia. However, statins cause side effects such as new-onset diabetes mellitus
402 (DM). The polyphenolic extract of chañar fruits flour is a natural source of HMG-CoA reductase
403 inhibitors that was able to suppress HMG-CoA reductase activity with a low IC_{50} value of 6 μ g
404 GAE/mL, Table 1. Phenolic compounds prevent *de novo* synthesis of cholesterol in the liver via the
405 suppression of HMG-CoA reductase. According to these results, we suggest that chañar
406 polyphenols may help reducing blood cholesterol and triglycerides by inhibition of lipid digestion
407 and absorption as well as by the inhibition of the cholesterol biosynthesis.

408 ***3.1.3. Effect of polyphenolic extract after treatment with digestive enzymes***

409 Since a large part of nutrient and non-nutrients are gastro-sensitive, the effect of gastroduodenal
410 digestion on bioactive compounds was studied. The polyphenolic extract after treatment with
411 digestive enzymes is able to inhibit enzymes such as α glucosidase and lipase with similar potency
412 to polyphenolic extract without digestion and consequently decreasing the bioavailability of food
413 nutrients (IC_{50} of 0.80 ± 0.05 and 4.50 ± 0.20 μ g GAE/mL, respectively). Furthermore, the

414 polyphenolic extract after treatment was active on α -amylase (IC_{50} 58 ± 2 μ g GAE/mL) with lower
415 potency than undigested extract (IC_{50} 25 ± 1 μ g GAE/mL).

416 **3.1.4. Inhibition of pro-inflammatory enzymes**

417 The effect of chañar fruit flour polyphenols extract was measured against the pro-inflammatory
418 enzymes COX, LOX and sPLA₂ (Table 1). Products of these enzymes (PGs, leukotrienes and
419 hydroperoxide) are important mediators of inflammation. The extract showed an IC_{50} value of 124
420 μ g GAE/mL against COX-2 while the reference drug nimesulide presented an IC_{50} of 0.39 μ g/mL.
421 The IC_{50} value of the extract for LOX was 48 μ g/mL with naproxen (selective inhibitor of LOX)
422 presenting an IC_{50} value of 14 μ g/mL. The extract was also active on sPLA₂ with an IC_{50} value of
423 225 μ g GAE/mL. Therefore, the polyphenolic extract of chañar fruits was able to inhibit the three
424 pro-inflammatory enzymes while other Argentinian fruits such as mistol or algarrobo were
425 selective inhibitors of LOX or COX, respectively (Cardozo et al., 2010; Pérez, et al., 2014). These
426 findings, together with the antioxidant activity observed in the polyphenolic extract of chañar
427 suggest that its consumption may contribute to the reduction of inflammation and could prevent or
428 ameliorate oxidative stress related diseases.

429 **3.2. Effect of polyphenolic extracts on oxidative stress**

430 Four basic mechanisms of antioxidant action have been described and are applicable to
431 polyphenolic compounds: (1) free radical scavenging activity, (2) quenching of singlet oxygen, (3)
432 chelating of transition metals, and (4) inhibition of free radical producing enzymes. The antioxidant
433 capacity of polyphenolic-enriched chañar fruits flour extract was determined using five different
434 experimental models. The polyphenolic extract exhibited effect as ABTS, hydroxyl radical and
435 H₂O₂ scavenger (Table 2), with higher antioxidant potency as HO[•] scavenger. Furthermore, the
436 extract showed electron or hydrogen donor capacity on ABTS (Table 2). In all experiments, the

437 polyphenols-enriched extract showed a dose-response relation with antioxidant capacity ($R^2 =$
438 0.998, 0.989, 0.977 and 0.964 for ABTS, H_2O_2 , HO^\bullet and β -carotene assays, respectively (Table 2).
439 Similar antioxidant activity on ABTS radicals was reported for the fruit of *Ziziphus mistol* ($SC_{50} =$
440 7.38 μ g GAE/mL) and *Prosopis* species (SC_{50} values between 7 and 29 μ g GAE/mL), native food
441 plants that grows in northern Argentina (Cardozo, Ordoñez, Alberto, Zampini, & Isla, 2011; Pérez
442 et al, 2014). Chañar extract was 28-folds more active (SC_{50} : 0.3 μ g GAE/mL) than mistol as
443 hydroxyl radical scavenger (SC_{50} : 14.13 μ g GAE/mL). Mistol ethanolic extract was also able to
444 protect linoleic acid from oxidation in the β -carotene system with IC_{50} value of 10.87 μ g GAE/mL,
445 with better activity than chañar.

446 3.3. Mutagenicity

447 In light of the potential nutritional and functional applications of chañar fruits flour, it is important
448 to prove the safety of products obtained from this fruit. The current study reports the results of *in*
449 *vitro* mutagenicity studies. In the Ames test, both on TA98 and TA100 *S. typhimurium* strains,
450 different doses of phenolic-enriched extract did not changed significantly the mutation frequencies
451 when compared to spontaneous ones, either in the presence or absence of metabolic activation (S9
452 mixture). The results indicate that *G. decorticans* fruit polyphenolic-enriched extract did not contain
453 compounds that could cause base substitution (detected in TA100) and frameshift (detected in
454 TA98) mutations. Furthermore, the extracts did not show the presence of pro-mutagenic
455 compounds. The absence of such an effect by *G. decorticans* fruits against *S. typhimurium* bacterial
456 strains is a positive step towards determining its safe traditional use. Taking together, the promising
457 chemopreventive activity and lack of mutagenic effect of chañar polyphenols in bacterial systems is
458 highly relevant for a possible agroindustrial development of this native fruit.

459 3.4. Identification of polyphenolics

460 The polyphenolic extract was fractionated by Sephadex LH-20. From the most active antioxidant
461 fractions of *G. decorticans* fruits flour, 39 phenolics were tentatively identified by HPLC-MS/MSⁿ
462 (Table 3). The assignment was based on comparison with literature, interpretation of the mass
463 spectra and co-chromatography with standards when available. The identity of the sugars and the
464 exact placement of the carbohydrate moieties in the aglycones remain to be established. HPLC
465 traces of the active fractions with the compound number are shown in Figure 2. The structure of the
466 compounds tentatively identified in the chañar flour polyphenolic fractions is presented in Figure 3.
467 Neutral loss scan experiments were carried out to identify the glycosides occurring in the samples.
468 The glycosides were assigned based on the neutral losses of 132, 146, 162 or 176 amu for a pentose,
469 rhamnose, hexose or glucuronic acid moieties from the M-1 ion, leading to the corresponding
470 aglycones (Barros, Dueñas, Ferreira, Carvalho, & Santos-Buelga, 2011; Simirgiotis, Theoduloz,
471 Caligari, & Schmeda-Hirschmann, 2009). The aglycones were identified by the [M-1] ions at 301
472 for quercetin, 315 for isorhamnetin, 285 for kaempferol and luteolin, 287 for the dihydro derivatives
473 eriodictyol and dihydro kaempferol, 269 for apigenin and 271 for naringenin. Further fragmentation
474 of the m/z ion of the aglycones yielded diagnostic MS³ fragments that allowed identification of the
475 genines. For quercetin, the ion at m/z 301 led to fragments at 179 and 151 amu, while isorhamnetin
476 yields fragments at m/z 300 and 151 from the [M-1] ion at 315 amu (Schieber, Keller, Streker,
477 Klaiber, & Carle, 2002). The differentiation of the genines kaempferol and luteolin, was based on
478 the MS³ experiments who allows a clear distinction based on the fragments of the aglycones (Fabre,
479 Rustan, de Hoffmann, & Quetin-Leclercq, 2001). While kaempferol give fragment ions of low
480 intensity, luteolin shows clear ions at m/z 241, 199 and 175 amu (Fabre et al., 2001). Flavanones
481 were assigned based on the work of Portet, Fabre, Rozenberg, Habib-Jiwan, Moulis, & Quetin-
482 Leclercq (2008) and Fabre et al. (2001). Eriodictyol fragments to a base ion of m/z 151 while

483 dihydrokaempferol shows the loss of water. The compounds tentatively identified comprised caffeic
484 acid glycosides, simple phenolics (protocatechuic acid and vanillic acid), a glycoside of vanillic
485 acid, *p*-coumaric acid and its phenethyl ester and several flavonoids. The flavonoids included free or
486 glycosylated flavonols (kaempferol, kaempferol methyl ether, quercetin, isorhamnetin), flavones
487 (luteolin and apigenin), dihydroflavones or flavanone (naringenin and eriodictyol) and
488 dihydroflavonols (dihydrokaempferol and taxifolin). Monoglycosides, diglycosides and
489 triglycosides occurs in the extract.

490 The main compounds were protocatechuic acid **4**, vanillic acid **6** and *p*-coumaric acid **9** in the
491 Fraction III, while Fraction IV afforded luteolin **26**, kaempferol rhamnoside hexoside **27**, apigenin
492 **35** and dihydroxy methoxy flavone **36** as main constituents. The sub-fractions were also analyzed
493 and all compounds identified in F-III y F-IV were confirmed in each sub-fraction. The sub-fraction
494 7 of F-IV yielded a mixture that was not found previously in F-IV : isorhamnetin **30**, isorhamnetin
495 rhamnoside hexoside **31**, dihydroxy methoxy flavone isopentyl and isoprenyl esters **38** and **39** as
496 accompanying compounds. The tentative identification and structure of the compounds is presented
497 in Table 3 and Figure 3, respectively. Four quercetin glycosides were identified, including a
498 hexoside, a pentosylhexoside, a rhamnosylhexoside and a dihexosiderhamnoside.
499 Isorhamnetinhexoside, hexoside pentoside, hexoside rhamnoside and rhamnosylhexoside occurs in
500 the complex mixtures. Kaempferol hexoside, rhamnosyl hexoside and hexosyl rhamnoside are
501 constituents of the extract as well as the rhamnosyl glucuronate of kaempferol methyl ether. The
502 flavone luteolin occurs as hexoside, hexosyl rhamnoside and rhamnosyl dihexoside while the
503 dihydro derivatives naringenin, eriodictyol and dihydrokaempferol were identified as
504 monohexosides. In the fruits, apigenin, luteolin, kaempferol, quercetin, isorhamnetin, and
505 dihydroxymethoxyflavone also occurs as aglycones.

506 From the flowers of chañar, Silva, López de Ruiz, & Ruiz, (2004) described the isolation and
507 identification of several flavonoid aglycones, including 3,3',4'-trihydroxyflavone, kaempferol,
508 quercetin, rhamnetin, isorhamnetin, morin, penduletin, jacein, jaceidin, patuletin, artemetin,
509 myricetin, naringenin, tricetin, diosmetin, zapotin, 5,7-dihydroxy-2'-methoxyflavone, apigenin,
510 baicalin, quercetagenin hexamethyl ether, gossypetin, quercetin 3',4',5,7-tetramethyl ether and
511 5,7-dihydroxy-4'-methoxyflavone. From the stem bark of chañar collected in the Bolivian Chaco,
512 Vila, Balderrama, Bravo, Almanza, Codina, Bastida, Connolly (1998) reported the isolation of
513 several isoflavanones, including (3R)-5,7,2',3'-tetrahydroxy-4'-methoxy-5'-prenylisoflavanone;
514 (3R)-7,2',3'-trihydroxy-4'-methoxy-5'-prenylisoflavanone and (3S)-3,7,2',3'-tetrahydroxy-4'-
515 methoxy-5'-prenylisoflavanone. Our finding shows a complex mixture of phenolic acids and
516 flavonoids differing in the oxidation patterns, number and placement of the sugar moieties. Further
517 work should be undertaken to fully characterize the compounds and to establish the chemical
518 variability in different populations of this valuable food resource of arid environments.

519 The content of major component of F- III such as protocatechuic acid (PCA), vanillic acid (VA) and
520 p-coumaric acid could be responsible for the effect of chañar flour extract on enzymes involved in
521 metabolic syndrome, mainly α -glucosidase and α -amilase (D'Archivio, Sczzocchio, Giovannini,
522 & Masella, 2014, Xiao, Chen, & Cao, 2014). It has been shown that the beneficial effects of PCA
523 and VA are mostly associated with its antioxidant as well as antihyperglycemic activities
524 (D'Archivio et al., 2014). PCA and VA have a key role in counteracting inflammation, due to its
525 ability to inhibit the synthesis and/or activity of most inflammatory mediators and regulatory
526 pathways involved in inflammation (D'Archivio et al., 2014). Furthermore, both acids seems to
527 have chemopreventive potential because they inhibits in vitro chemical carcinogenesis and exerts

528 pro-apoptotic or antiproliferative effects in different tissues (Tanaka, Tanaka, & Tanaka, 2011;
529 D'Archivio et al., 2014).

530 In relation to flavonoids of chañar flour, flavones from F-IV such as luteolin and apigenin and its
531 derivatives exhibits antioxidant and anti-inflammatory activities mainly as potential inhibitors of
532 COX-2 (González-Castejón & Rodríguez-Casado, 2011; Ya-Di et al., 2011). Luteolin inhibited α -
533 glucosidase in a noncompetitive mechanism and the luteolin-enzyme binding was driven mainly by
534 hydrophobic interactions with conformational changes of the enzyme. The luteolin had a high
535 affinity close to the active site pocket of α -glucosidase and indirectly inhibited the catalytic activity
536 of the enzyme. Recently it was also reported that the glycosylation of flavonoids lowered the
537 inhibition against α -glucosidase depending on the conjugation position and the class of sugar
538 moieties. The decreased inhibitory effect against α -glucosidase after glycosylation may be due to
539 the increasing molecular size and polarity, and the non-planar structure. When a hydroxyl moiety is
540 substituted by a glycoside, the steric hindrance may happen, which weakens the binding interaction
541 between flavonoids and α -glucosidase (Xiao et al., 2014).

542 Flavonols such as kaempferol and quercetin and its derivatives were found to be especially effective
543 lipoxygenase inhibitors while quercetin was found to be an effective inhibitor of PLA2 in human
544 leukocytes (Kim et al., 2014).

545 The hypocholesterolemic activity of polyphenolic extract of chañar flour could be related to the
546 presence of flavanone such as naringenin and eriodictiol. In previous report was demonstrated that
547 the statin-like flavanones extracted from bergamot peel, exert a similar behavior respect to
548 commercial simvastatin on a model of hypercholesterolaemic rats (Di Donna et al., 2014). Other
549 studies showed eriodictiol as an inhibitor of rat platelet 5- and 12-lipoxygenases which are
550 involved in the biosynthesis of several bioregulators that are closely related to the pathogenesis of

551 diseases such as allergy, atherosclerosis and cancer (González-Molina, Domínguez-Perles, Moreno,
552 & García-Viguera, 2010). Flavanones show strong antioxidant and radical scavenging activity,
553 antiviral, antimicrobial activities, beneficial effects on capillary fragility, and an ability to inhibit
554 human platelet aggregation as well as anti-ulcer properties (Tomás-Navarro, Vallejo, & Tomás-
555 Barberán, 2014).

556 **4. Conclusion**

557 The chañar fruits have been a relevant food source for the South American amerindians since pre-
558 hispanic times. Different ethnic groups of Argentina consider the chañar tree as a gift from God and
559 are cared for by the Pachamama (mother earth). On the other hand, are considered as health- and
560 energy-giving trees. At present, the sweet and pleasant tasting fruits are still consumed raw or
561 processed into several products used in the local cuisine. The ripe fruits flour contains a complex
562 mixture of polyphenols (phenolic acids and flavonoids) that present relevant functional properties.
563 These findings further support the idea that a diet including chañar fruits flour or chañar fruits, may
564 be preventive against diseases that are associated with oxidative stress, inflammatory mediators and
565 metabolic syndrome. Our results on the functional properties of chañar fruits flour encourages
566 further clinical studies to determine optimal dietary regimens to achieve the desired beneficial
567 health effects. In addition, components of chañar fruits flour are attractive targets for the scientific
568 community to develop novel food products for treatment/prevention of these life-threatening
569 diseases. The nutritional and functional properties that were demonstrated for the chañar fruits flour
570 could stimulate the preservation of chañar forests including sustainable development and
571 management for commercial purposes, contributing to regional development in arid zones.

572

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ACCEPTED MANUSCRIPT

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Figures

Figure 1: *Geoffroea decorticans* tree growing in the Provincia de Tucumán, Argentina. A: Detail of the tree trunk and bark; B: Tree in the “parque chaqueño” formation; C: Ripe fruits in the tree; D: Ripe fruits are ready to be collected when they fall from the tree.

Figure 2: HPLC chromatograms of polyphenolic extract from *Geoffroea decorticans* fruits

Ultraviolet (UV) detection at 254 nm. A) Fraction III; B) Fraction IV; C) Sub fraction -7.

Figure 3: Compounds tentatively identified in chañar fruits. *Identified by comparison with standards.

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Figure 1:

A



B



C

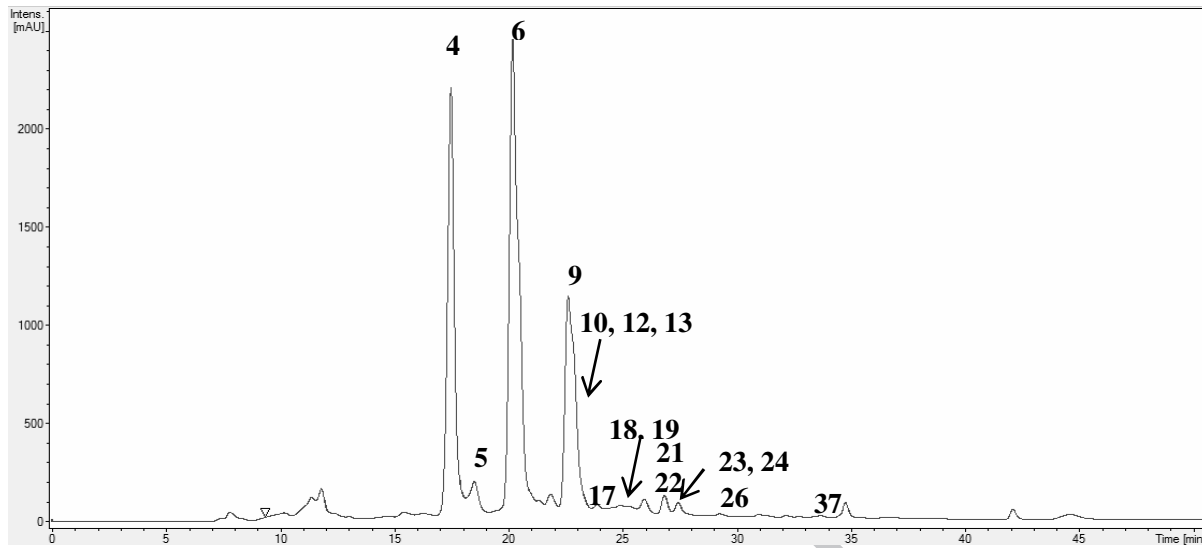


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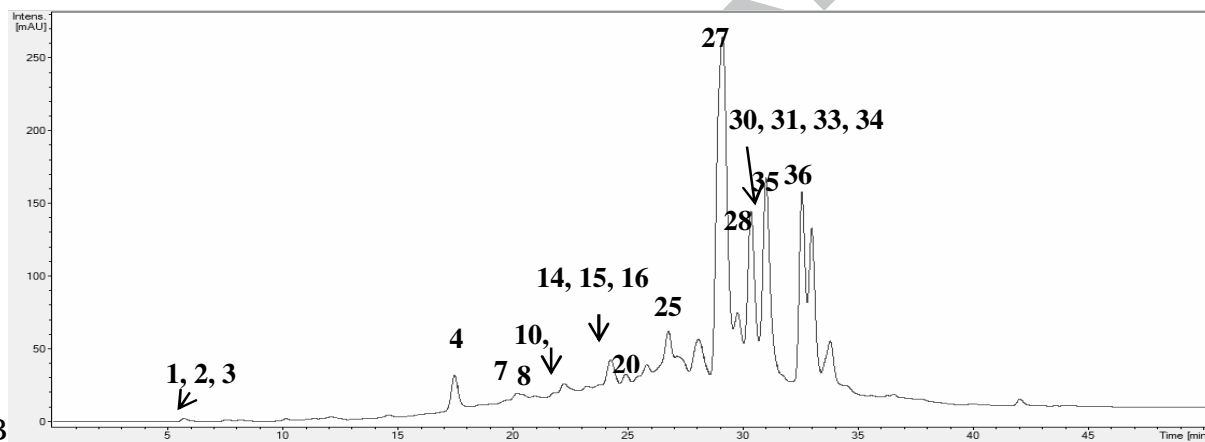


Figure 2:

A



B



C

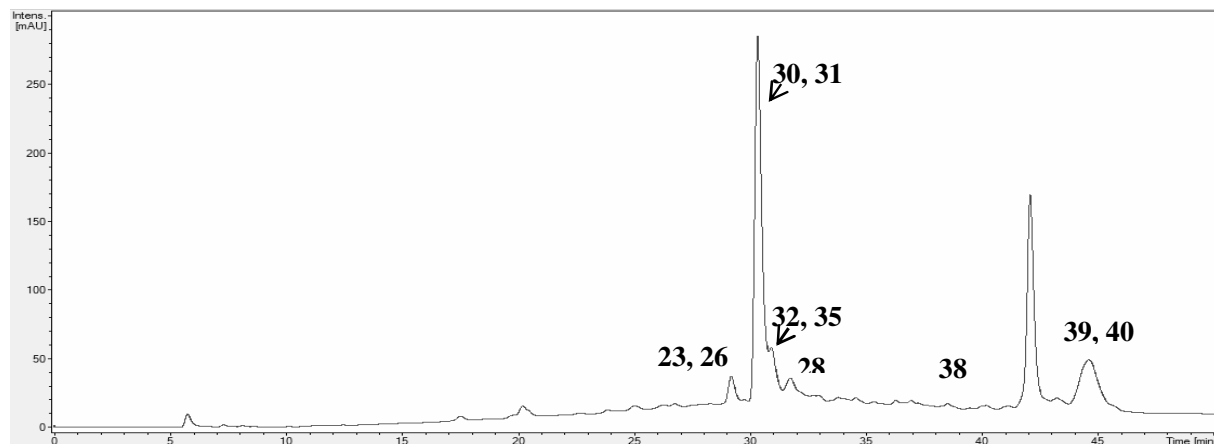
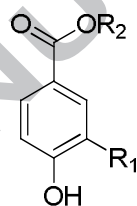
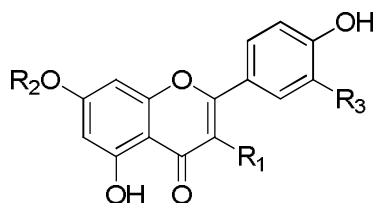
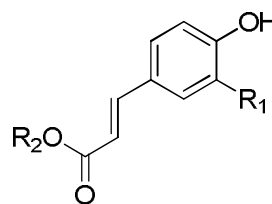


Figure 3;



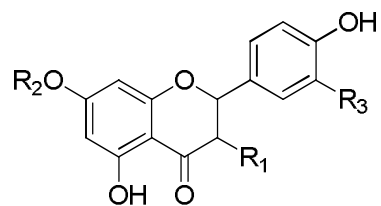
	R ₁	R ₂	
4	OH	H	Protocatechuic acid*
5	OCH ₃	Hexose	
6	OCH ₃	H	Vanillic acid*

	R ₁	R ₂	R ₃	
35	H	H	H	Apigenin*
32	H	Dirhamnoside	H	
26	H	H	OH	Luteolin
22	H	Hexose	OH	
23	H	Hexose-rhamnose	OH	
17	H	Rhamnose-dihexose	OH	



	R ₁	R ₂	R ₃		R ₁	R ₂	
28	OH	H	H	Kaempferol*	OH	H	Caffeic acid
20	Hexoside		H		1	Hexose-rhamnose	
19	Hexoside-rhamnoside		H		2	Hexose	
18	Hexoside-pentoside		H		3	Dihexose	
27	Rhamnose-hexose		H		9	H	<i>p</i> -Coumaric acid*
29	OH	CH ₃	H		37	Phenethyl	
25	Rhamnose-glucuronic acid						
33	Dirhamnoside						
34	Rhamnose-glucuronic acid						

	R ₁	R ₂	R ₃	
	OH	H	OH	Quercetin*
15	Hexose		OH	
10	Hexose-pentose		OH	
14	Rhamnose-hexose		OH	
12	Dihexose-rhamnose		OH	
30	OH	H	OCH ₃	Isorhamnetin
11	Hexose		OCH ₃	
13	Hexose-pentose		OCH ₃	
21	Hexose-rhamnose		OCH ₃	
24	Prenyl, hexose		OCH ₃	
31	Rhamnose-hexose		OCH ₃	



	R ₁	R ₂	R ₃	
	H	H	H	Naringenin
8	H	Hexose	H	
	H	H	OH	Eriodictyol
16	H	Hexose	OH	
	OH	H	H	Dihydrokaempferol
7	OHexose	H	H	

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Table 1: Effect of polyphenols enriched extract of Chañar flour and reference compounds on Enzymes related to carbohydrate metabolism, fat metabolism and inflammatory processes. Results are reported as IC₅₀ values in µg GAE/mL.

Enzyme	Polyphenols enriched extract IC ₅₀ values (µg GAE/mL)	Reference compound	Reference compounds IC ₅₀ values (µg/mL)
Enzymes related to carbohydrate metabolism			
α amylase	25.00±1.00 ^b	Acarbose	1.25±0.10
α-glucosidase	0.68±0.05 ^a	Acarbose	25.00 ± 1.00
Enzymes related to fat metabolism			
Lipase	4.00±0.20 ^a	Orlistat	0.08±0.01
Hydroxymethyl glutaryl CoA reductase	6.00±1.00 ^a	Pravastatin (SIGMA 15909)	1µl inhibition 18%
Enzymes related to inflammatory processes			
COX-2	124±5 ^b	Nimesulide	0.39±0.10
LOX	48±2 ^a	Naproxen	14.00 ±1.00
PLA2	225±5 ^c	Acetylsalicylic acid	65.00± 1.00

Different letters (a, b, c) in the same column in each biological assay show significant differences among effect of polyphenols on enzyme activity according to Tukey's test ($p \leq 0.05$).

Table 2: Antioxidant activity of polyphenols enriched extract of Chañar flour. Results are presented as SC₅₀ or IC₅₀ in µg GAE/mL

Assay	Polyphenols enriched extract	Reference compound	Reference compounds IC ₅₀ values (µg/mL)
ABTS ^{•+} (SC ₅₀)	2.8±0.2 ^b	Quercetin	6.7±0.3
		BHT	7.7±0.4
H ₂ O ₂ (SC ₅₀)	23.0±1.0 ^c	Quercetin	12.0±1.0
HO [•] (SC ₅₀)	0.30±0.05 ^a	Quercetin	30.0±2.0
% Lipid protection β-Carotene (IC ₅₀)	22.0±1.0 ^c	Quercetin	9.8 ±0.9
		BHT	3.9 ±0.2
BSA protection (50% protection)	18.0±1.0 ^c	Quercetin	15.0±1.0

Different letters (a, b, c) in the same column show significant differences among antioxidant effect of polyphenols by different mechanisms according to Tukey's test ($p \leq 0.05$).

Table 3. Tentative identification of phenolic compounds in the most active antioxidant fractions of chañar fruit

	Rt (min)	[M-H] ⁻	MS/MS	Tentative identification
1	6.2	487	341, 179, 161	Caffeic acid hexosiderhamnoside
2	6.3	683	341 (2M-1)	Caffeic acid hexoside
3	6.6	503	341	Caffeic acid dihexoside
4	16.7-18.0	153	109	Protocatechuic acid
5	18.0-19.1	329	167	Vanillic acid hexoside
6	19.1-21.3	167	152, 123, 107	Vanillic acid
7	20.3	449	287	Dihydrokaempferolhexoside
8	20.6	433	271, 151	Naringeninhexoside
9	22.0-23.9	163	119	<i>p</i> -coumaric acid
10	22.7-23.0	595	301, 179, 151	Q-hexosidepentoside
11	22.9	477	315	Isorhamnetin/Rh hexoside
12	23.1-23.2	771	301, 179, 151	Q-dihexosiderhamnoside
13	23.2	609	477, 315, 300	Isorhamnetin/Rh hexosidepentoside
14	23.5	609	301, 179, 151	Q-rhamnosidehexoside
15	23.5	463	301, 179, 151	Q hexoside
16	23.6-23.9	449	287, 151	Eriodictyolhexoside
17	23.8	755	609, 285	Luteolinrhamnosidedihexoside
18	24.0	579	447, 285	Kaempferolhexosidepentoside
19	24.7	593	447, 285	Kaempferolhexosiderhamnoside
20	24.8	447	285	K hexoside
21	24.9-25.0	623	477, 315, 207	Isorhamnetin/Rh hexosiderhamnoside
22	26.3	447	285	Luteolinhexoside
23	26.8-29.5	593	285	Luteolin hexose rhamnose
24	27.6	545	477, 315, 300	Prenylisorhamnetinhexoside
25	28.3	621	299, 284	K methyl ether rhamnosideglucuronate
26	28.6-28.7	285	267, 241, 217, 199, 175	Luteolin
27	28.7	593	285	K rhamnosidehexoside
28	29.7-31.7	285	257, 239, 199, 119	Kaempferol
29	29.7	299	284, 151	Kaempferol methyl ether
30	29.7	315	300	Isorhamnetin/Rhamnetin
31	29.9	623	315, 299, 285, 236	Isorhamnetin/Rhrhamnosidehexoside
32	30.4	561	269	Apigenindirhamnoside
33	30.4	591	299, 269	K methyl ether dirhamnoside
34	30.4	621	299	K methyl ether rhamnosideglucuronate
35	30.2-30.4	269	225, 177, 149, 133	Apigenin
36	32.4	283	268	Dihydroxymethoxy flavoneGenkwanin
37	32.9	267	163, 119	<i>p</i> -coumaric acid phenethyl ester
38	38.5	367	283, 176	Dihydroxymethoxyflavoneisopentyl ester

39	43.0-44.1	365	283	Dihydroxymethoxyflavoneisoprenyl ester
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The *Geoffroea decorticans* fruits (chañar) are consumed in Argentina as flour, arropo or beverage

39 phenolic compounds were tentatively identified in the polyphenolic extract of chañar flour

Polyphenols from *chañar* fruits flour affects enzymes involved in metabolic syndrome

Polyphenols from *chañar* fruits flour exhibited antioxidant and anti-inflammatory activities

Chañar flour could be used as functional food alone or combined with other flours

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